

Chapter 5: Work and Machines Guided Reading

Read the indicated sections and answer the following questions. They are IN ORDER from the text.

Pages 126-131

- How are an applied force and an object's motion related when work is done? The applied force and the object's motion must be in the same direction.
- Using figure 3- Explain how the energy of the box changes as the student climbs the stairs? Potential energy increases because the box is being moved up the stairs.
- Work equation. Label and include the units for each variable.
 $W = Fd$ $W = \text{work (J)}$ $F = \text{force (N)}$ $d = \text{distance (m)}$
- How is power related to work? Power is the amount of work is done per second.
- Power equation. Label and include the units for each variable.
 $P = W/t$ $P = \text{power (W)}$ $W = \text{work (J)}$ $t = \text{time (s)}$

Pages 132-137

- How does a ramp make lifting an object easier? Less force is needed to lift an object when using a ramp.
- Using figure 10- Describe how the hammer changes the input force. The hammer produces an output force that is greater than the input force.
- Mechanical Advantage equation. Label and include the units for each variable.
 $MA = \text{Output force} / \text{Input force}$ $MA = \text{mechanical advantage (no units)}$ Both forces are in N
- Why is the output work always less than the input work for a real machine? Because some of the input force is converted into thermal energy.
- Efficiency equation. Label and include the units for each variable
 $\text{Efficiency} = \text{Output work (J)} / \text{Input work (J)} \times 100$

Pages 138-146

- Identify each of the classes of levers and give an example of each. 1st- see saw 2nd- wheel barrow 3rd- crowbar
- Identify each of the types of pulleys and give an example of each. Fixed pulley- well Movable pulley- static line Block & tackle- used on sailboats.
- How does a movable pulley reduce the input force needed to lift a weight? You have to increase the distance the rope moves by double.
- Give an example of a wheel and axle, inclined plane, screw, and wedge. Wheel & axle- screwdriver Inclined plane- stairs Screw- bolt Wedge- doorstop
- Define compound machine. Give an example of one and list the types of machines it includes.
Combination of 2 or more simple machines.

Guided Reading Study Work Chemistry Answers Chapter 5

Peter L. Kresan, Reed Mencke



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Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e

Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry [A Practical Guide to Scientific Writing in Chemistry](#) Andrew Terhemen Tyowua, 2023-03-31 Successful completion of postgraduate studies especially PhD and career advancement in academia strongly depend on the ability to publish scientific papers or books and attract research grants However many chemical scientists find preparing scientific papers and research grant and book proposals difficult partly because of insufficient training in writing and partly because there are few practical books to enable them to learn the art This step by step practical guide is intended mainly for postgraduate students and early career researchers in chemical science and the libraries that serve them but will also be useful to other scientists Key Features Improves the reader's chances of getting their manuscript published in chemistry journals Increases the likelihood of winning research grants in chemistry Takes a lead by the hand approach Contains chapters on the preparation of graphical abstracts and research highlights Uses sketches and other illustration styles to aid mental visualization of concepts Contains practical examples taken from published papers and successful research grant proposals *Student Study Guide/solutions Manual to Accompany Foundations of Inorganic, Organic, and Biological Chemistry* Katherine J. Denniston, 1995 This student study guide has been written to accompany Foundations of Inorganic Organic and Biological Chemistry by Caret Denniston and Topping It was designed to complement the text not to be used in place of the text Each chapter of the study guide contains the following sections reorganized and expanded set of learning goals concise chapter summary in chapter solved problems list of key terms review problems Preface **Addison-Wesley Chemistry** Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L. Waterman, Prentice-Hall Staff, 2001-02 To purchase or download a workbook click on the Purchase or Download button to the left To purchase a workbook enter the desired quantity and click Add to Cart To download a free workbook right click the FREE Download PDF link and save to your computer This will result in a faster download as opposed to left clicking and opening the link *Organic Chemistry, 13e Student Study Guide and Solutions Manual* T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-05-03 Organic Chemistry Student Study Guide and Solutions Manual 13th Edition offers the full solutions for select exercises from the text **The Official SAT Study Guide, 2018 Edition** The College Board, 2017-05-16 Includes 8 real SATs and official answer explanations Cover [The Logic of Biochemical Sequencing](#) D. Blackman, 2018-01-18 The Logic of Biochemical Sequencing examines how to determine the primary structures of proteins and DNA and use them to stimulate the process of logical problem solving It concentrates on

sequencing work and stresses the thought processes needed to make sense of what might otherwise be indecipherable data. The book also introduces biocryptography which serves as a basis for four short stories that use the results of sequence determinations to provide clues to higher order problems. Problems in the book range from elementary to difficult and solutions to all problems are provided many of them completely worked out. The book is an excellent supplementary text for students in a full year biochemistry course as well as for biochemists and molecular biologists. *Study Guide & Solutions Manual to Accompany Organic Chemistry, Third Edition* G. Marc Loudon, Joseph G. Stowell, 1995 Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual T. W. Graham Solomons, Craig B. Fryhle, Scott A.

Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition. The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond. A central theme of the authors' approach to organic chemistry is to emphasize the relationship between structure and reactivity. To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms. The authors' philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters. The structural aspects of the authors' approach show students what organic chemistry is. Mechanistic aspects of their approach show students how it works. And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us.

Understanding Earth Student Study Guide Peter L. Kresan, Reed Mencke, 2006-05-03 The guide helps students prepare for lectures and exams with a heavy emphasis on utilizing the book's Web resources. Resources in Education, 2000-04 **Organic Chemistry** Michael B. Smith, 2016-03-09 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization. Using several techniques to develop a relational understanding it helps students fully grasp the essential concepts at the root of organic chemistry. This new edition was rewritten largely with the feedback of students in mind and is also based on the author's classroom experiences using the first edition. Highlights of the Second Edition include Reorganized chapters that improve the presentation of material. Coverage of new topics such as green chemistry. Adding photographs to the lectures to illustrate and emphasize important concepts. A downloadable solutions manual. The second edition of Organic Chemistry An Acid Base Approach constitutes a significant improvement upon a unique introductory technique to organic chemistry. The reactions and mechanisms it covers are the most fundamental concepts in organic chemistry that are applied to industry biological chemistry biochemistry molecular biology and pharmacy. Using an illustrated conceptual approach rather than presenting sets of principles and theories to memorize it gives students a more concrete understanding of the material. *The*

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Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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