

Stoichiometry Guided Practice:

Name _____

Period ____ Date ____/____/____

Molar to Molar Stoichiometry:



a. What is the mole ratio of CaSO_3 to water?

b. How many moles of CaSO_3 will be produced if 4 moles of water are also produced?

c. What is the mole ratio of Ca(OH)_2 to water?

d. How many moles of water are produced if 4 moles of Ca(OH)_2 are used in the reaction?



a. What is the mole ratio of water to Ca_2CO_3 ?

b. How many moles of H_2O will be produced if 0.2000 moles of Ca_2CO_3 are produced?

c. What is the mole ratio of Ca_2CO_3 to NH_3 ?

d. How many moles of NH_3 will be produced if 0.5 moles of Ca_2CO_3 are used in the reaction?



a. What is the mole ratio of Cl_2 to PCl_3 ?

b. How many moles of Cl_2 will react with 0.4624 moles of PCl_5 ?

Guided Study Work Answers Stoichiometry

Margaret G. Kimble



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