

Here are the four formulas being used as examples:

Molecular Formula	Empirical Formula
H_2O	H_2O
$\text{C}_2\text{H}_4\text{O}_2$	CH_2O
CH_2O	CH_2O
$\text{C}_6\text{H}_{12}\text{O}_6$	CH_2O

Notice two things:

1. The molecular formula and the empirical formula **can be identical**.
2. You scale up from the empirical formula to the molecular formula by a whole number factor.

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