

CHAPTER 21 REVIEW

Nuclear Chemistry**Teacher Notes and Answers****Chapter 21****SECTION 1****SHORT ANSWER**

1. b
2. a
3. d
4. b
5. b
6. a. helium-3
b. helium-4
7. 13
8. atomic number
9. a. True
b. atom Z
c. atom Z
10. a. $^{14}_7\text{N}$
b. $^{60}_{28}\text{Ni}$
11. $E = mc^2$
12. a. Iron-56 has 26 protons; carbon-14 has 6 protons.
b. Iron-56 has 30 neutrons; carbon-14 has 8 neutrons.
c. Iron-56 is likely to be stable.
13. 0.172 46 amu

Section 2**SHORT ANSWER**

1. a
2. d
3. a
4. a. c
b. a
c. d
d. b
5. a. False
b. True
c. True
d. False
6. $^{238}_{92}\text{U}$
7. $^{210}_{82}\text{Pb} \rightarrow ^0_{-1}\beta + ^{210}_{83}\text{Bi}$
8. The value of x is 5. The total mass of the reactants is 252 amu, therefore the total mass of the products must be 252. Then

252 – 247 leaves 5 amu for neutrons at 1 amu each.

9. 42.9 days
10. 5.0 mg
11. 0.24 mg

Section 3**SHORT ANSWER**

1. d
2. c
3. b
4. a. c
b. b
c. a
d. d
5. Alpha radiation penetrates the least and is most easily absorbed.
6. The greater the ratio of lead-206 to uranium-235 in the sample, the older the rock sample is.
7. 6.2 mg
8. 3.5 units

Section 4**SHORT ANSWER**

1. a. 1
b. 3
c. 1
d. 2
e. 3
2. a. a, c
b. d
c. a
d. b
3. a. c
b. a
c. f
d. b
e. e
f. d
4. b
5. The ratio of He atoms to H atoms in a star's composition becomes larger. As a star ages, H atoms fuse into He atoms. A new star is almost 100% H, but the older the star

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