

# Concept Review

## Section: What Is Matter?

1. **Classify** the following as a homogeneous, *O*, or a heterogeneous, *E*, mixture.

\_\_\_\_\_ **a.** a pail of sand and water

\_\_\_\_\_ **b.** air

\_\_\_\_\_ **c.** human blood

\_\_\_\_\_ **d.** banana split

\_\_\_\_\_ **e.** chocolate syrup

\_\_\_\_\_ **f.** sea water

2. **Compare and contrast** atoms and molecules.

\_\_\_\_\_  
\_\_\_\_\_

3. **Explain** the difference between a pure substance and a homogeneous mixture.

\_\_\_\_\_  
\_\_\_\_\_

4. **Classify** each of the following as an element or a compound.

\_\_\_\_\_ **a.** benzene,  $C_6H_6$

\_\_\_\_\_ **b.** aluminum, Al

\_\_\_\_\_ **c.** aspirin,  $C_9H_8O_4$

\_\_\_\_\_ **d.** titanium, Ti

\_\_\_\_\_ **e.** acetylene,  $C_2H_2$

\_\_\_\_\_ **f.** zinc, Zn

5. **Explain** why elements and compounds are pure substances.

\_\_\_\_\_  
\_\_\_\_\_

6. **Determine** which of the following are pure substances and which are mixtures.

\_\_\_\_\_ **a.** salt water

\_\_\_\_\_ **b.** isopropyl alcohol,  $C_3H_8O$

\_\_\_\_\_ **c.** mercury, Hg

\_\_\_\_\_ **d.** ammonia,  $NH_3$

\_\_\_\_\_ **e.** an egg yolk

\_\_\_\_\_ **f.** honey

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**Lei Huang**

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