



Example 5

Draw the Lewis diagram for each molecule.

1. N_2
2. CH_2O (The carbon atom is the central atom.)

Solution

1. The bond between the two nitrogen atoms is a triple bond. The Lewis diagram for N_2 is as follows:



2. In CH_2O , the central atom is surrounded by two different types of atoms. The Lewis diagram that fills each atom's valence electron shell is as follows:



Note

One application of CH_2O , also called formaldehyde, is the preservation of biological specimens. Aqueous solutions of CH_2O are called formalin and have a sharp, characteristic (pungent) odor.

Skill-Building Exercise

Draw the Lewis diagram for each molecule.

1. O_3
2. C_2H_2

Concept Review Exercise

1. What is one clue that a molecule has a multiple bond?

Answer

1. If single bonds between all atoms do not give all atoms (except hydrogen) an octet, multiple covalent bonds may be present.

Key Takeaway

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Patrick Vollmar

A red circular graphic with a gradient, appearing as a partial circle or a thick arrow pointing to the right, located on the right side of the light blue bar.

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