A Company of the Comp	Professional	District

Physical Science — 2nd Semester — Final Exam Study Guide — (2012-2013)

Physical Science Basics

What tooks would you want to use to find the:

Mass of an object? triple beam balance Barrier Strictor's No.

Volume of an object? rules or water displacement method Restic St Unit? and or red.

Length of an object? Ruler

Stack: S1 Unit? meter Density of an object? graduated cylinder Blooks St Unit 7 mil.

What is the formula for calculating density? resun/volume.

What is the density of a cube that is 35 g and has a volume of 3 cm 7 Se/cm

Compare the density of solid NLO to liquid NLO, and explain what observable effect this has.

solid HyDis less dense than liquid HyD.

Chemistry Basics.

Dernet (def.) substance made up of one kind of store

Atom (def.) basic building block of matter.

Composed (def.) a majerate that centains atoms or molecules of two or more different elements

4. Molecule (4ef.) a combination of two at more of the same or different atoms that are bonded together

Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that reary scientists throughout history have created anodels to describe them.

Why have storaic theories channel over time? Better technology come around.

Solumber	Year	Major Discovery/ Contribution	Morne of atomic model
John Deltse	1862	Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different mesors.	Billiard Boll Model
J.J. Thoreson	1887	Negatively charged electron	Plum Pudding Model
Niets Bohr	1909	Electrons move fined arbits around the atom's nucleus in specific layers, or siectron shalls. Atoms absorb or give off energy when the electrons move from one shall (grappy level) to enother	Stohr Model
orwest Muldherford	1900	1909 (Based on lifes that positive reflects positive) atom mostly empty space; electrons orbit rendemly around positively charged reciess	Planetary Model

4. Previouslie Testates

The current periodic table is arranged by increasing atomic number.

Reackes the Periodic Table

- An element's properties can be predicted from its location in the periodic table.
- The atomic number represents the number of pretent.
- The atomic mass represents the number of pentrons and posture

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Michael Seilmaier

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Chemical News and Journal of Physical Science William Crookes, James H. Gardiner, Gerald Druce, H. W. Blood-Rvan, 1919 The Publishers' Trade List Annual ,1886 **Resources for Teaching Middle School Science** Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-03-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance

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Quarterly ,1928 The Science Teacher ,1998 Some issues are accompanied by a CD ROM on a selected topic

Resources in Education ,2000-10 Textbooks in Print ,1957 Handbook on Contemporary Education Steven

E. Goodman, 1976

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