Chemistry (B) HW Chapter 13 (April 1 Due April 16 Test April 17)

Matching

	Match each item with the correct	statement below.	
	a. melting point	d. evaporation	
	 b. boiling point 	e. vaporization	
	c. sublimation	f. normal boiling point	
	1. vaporization at the surface of a liqui	d that is not boiling	
2	the conversion of a liquid to a gas be	the conversion of a liquid to a gas below the boiling point	
3	 the temperature at which the vapor p 	the temperature at which the vapor pressure of a liquid is equal to the external pressure	
- 4	4. the temperature at which the vapor p	the temperature at which the vapor pressure of a liquid is equal to 1 atmosphere	
	the temperature at which a solid changes into a liquid		
	the change of a solid directly to a va	рог	
	Match each item with the correct	statement below.	
	a. kinetic theory	d. barometer	
	 atmospheric pressure 	e. kinetic energy	
	c. vapor pressure		
8	the energy an object has due to its motion		
	the pressure resulting from the collision of atoms and molecules with objects		
1.0	a measure of the force exerted by a gas above a liquid		
	Choice the choice that best completes the states 1. According to the kinetic theory, coll a. are perfectly elastic b. are inelastic		
12	Which of the following statements is part of the kinetic theory? a. The particles of a gas move independently of each other. b. The particles in a gas move rapidly. c. The particles in a gas are relatively far apart. d. all of the above		
13	The average speed of oxygen molecules in air is about		
	 0 km/h 	 e. 1700 km/h 	
	 b. 170 km/h 	d. 17,000 km/h	
14	Which of the following statements is NOT true, according to the kinetic theory? a. There is no attraction between particles of a gas. b. Only particles of matter in the gaseous state are in constant motion. c. The particles of a gas collide with each other and with other objects. d. All of the statements are true.		

Holt Chemistry Chapter 13 Test Answers

RJ Alexander

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