

Chapter 9 Study Guide

Holt Chemistry: Stoichiometry

Questions

1. What is stoichiometry? Finding mass of product formed (or reactant consumed) when given a mass of another species in the reaction, by using the mole to mole ratio from the balanced equation.
2. A balanced equation verifies the law of conservation of mass: Total mass of reactants must equal total mass of products.
3. Give the 3 steps of converting the mass (grams) of one substance to the mass (grams) of another substance: 1. Convert grams to moles using molar mass; 2. Mole to Mole ratio from balanced equation; 3. Convert moles back to grams using molar mass.

Holt Chemistry Stoichiometry Study Guide Answers

Olga I. Agapova



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Holt Chemistry Holt Rinehart & Winston, 2003-01-24 **Holt Chemistry** Ralph Thomas Myers, 2004 **The United States Catalog** Mary Burnham, Carol Hurd, 1928 **Children's Books in Print, 2007** ,2006 **Children's Books in Print** R R Bowker Publishing, Bowker, 1999-12 **Books in Print Supplement** ,2002 **Forthcoming Books** Rose Army, 2003-04 **Modern Chemistry** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **General Chemistry: Experiment and Theory** Richard H. Eastman, 1970 *ChemDiscovery Teacher Edition* Olga I. Agapova, 2002

Technology in the Curriculum: Science resource guide ,1986 **International Chemistry Directory** ,1969
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