

An isotope has a half life of 33 seconds. How long would it take for a 100g sample to get to 15g?

$$\text{remaining} = \frac{\text{initial}}{2^{\text{age}/\text{HL}}} \rightarrow 2^{\text{age}/\text{HL}} = \frac{\text{initial}}{\text{rem}} \quad \begin{matrix} 6^{\text{age}} = y \\ \log_6 y = x \end{matrix}$$

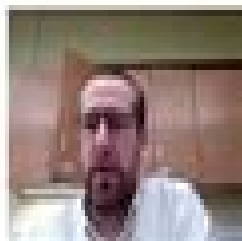
$$\log_2 \frac{\text{initial}}{\text{rem}} = \text{age}/\text{HL}$$

$$15\text{g} = \frac{100\text{g}}{2^{\text{age}/33\text{sec}}}$$

$$\log_2 \frac{100}{15} = \text{age}/\text{HL}$$

$$\log_2 6.67 = \text{age}/\text{HL}$$

$$\log 6.67 / \log 2 = \text{age}/$$



Half Life Algebra 2 Word Problems

Jicheng Xie



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