

## EXERCISE NO. 4

### DENSITY, SPECIFIC GRAVITY AND SPECIFIC VOLUME

#### I. INTENDED LEARNING OUTCOMES

At the end of the exercise, the students should be able to:

1. calculate density, specific gravity and specific volume of different liquids.
2. determine the specific gravity of unknown liquids using the data obtained from the specific gravity bottle (pycnometer) method.
3. calculate weight or volume of the substance given the specific gravity.

#### II. DISCUSSION

**Density** is defined as mass per unit volume of a substance at a fixed temperature and pressure. Density can be calculated by dividing the mass of a substance by its volume, expressed as:  $Density = \frac{Mass}{Volume}$ . The density of liquids is usually expressed as grams per milliliter (g/mL).

**Specific gravity** is the ratio of the weight of a given substance to the weight of an equal volume of water at a similar temperature, expressed as:

$$\text{Specific gravity} = \frac{\text{Weight of substance}}{\text{Weight of equal volume of water}}$$

In the U.S. Pharmacopeia, the standard temperature for specific gravities is 25°C, unless otherwise stated. Specific gravity is a dimensionless property and therefore has no units. The specific gravity of water is 1; therefore, the volume of water in milliliters is usually equal to its weight in grams. If the weight of a particular volume of a substance is known, its specific gravity can be calculated by dividing its weight by its volume:

$$\text{Specific gravity} = \frac{\text{Weight of substance (g)}}{\text{Volume of substance (mL)}}$$

This equation is similar to the one for density. So, an easy way to calculate specific gravity is to calculate the density and write the value without any units. If the specific gravity of a solution is known, interconversions between the volume of a substance and its weight are possible using the following expressions:

$$\text{Weight of substance (g)} = \text{Volume of substance (mL)} \times \text{Specific gravity}$$

$$\text{Volume of substance (mL)} = \frac{\text{Weight of substance (g)}}{\text{Specific gravity}}$$

## Gravity The Taking Volume

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Ephraim Lessing, Bjørnstjerne  
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### **Table of Contents Gravity The Taking Volume**

1. Understanding the eBook Gravity The Taking Volume
  - The Rise of Digital Reading Gravity The Taking Volume
  - Advantages of eBooks Over Traditional Books
2. Identifying Gravity The Taking Volume
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Gravity The Taking Volume
  - User-Friendly Interface
4. Exploring eBook Recommendations from Gravity The Taking Volume



- Personalized Recommendations
- Gravity The Taking Volume User Reviews and Ratings
- Gravity The Taking Volume and Bestseller Lists
- 5. Accessing Gravity The Taking Volume Free and Paid eBooks
  - Gravity The Taking Volume Public Domain eBooks
  - Gravity The Taking Volume eBook Subscription Services
  - Gravity The Taking Volume Budget-Friendly Options
- 6. Navigating Gravity The Taking Volume eBook Formats
  - ePub, PDF, MOBI, and More
  - Gravity The Taking Volume Compatibility with Devices
  - Gravity The Taking Volume Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Gravity The Taking Volume
  - Highlighting and Note-Taking Gravity The Taking Volume
  - Interactive Elements Gravity The Taking Volume
- 8. Staying Engaged with Gravity The Taking Volume
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Gravity The Taking Volume
- 9. Balancing eBooks and Physical Books Gravity The Taking Volume
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Gravity The Taking Volume
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Gravity The Taking Volume
  - Setting Reading Goals Gravity The Taking Volume
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gravity The Taking Volume

- Fact-Checking eBook Content of Gravity The Taking Volume
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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