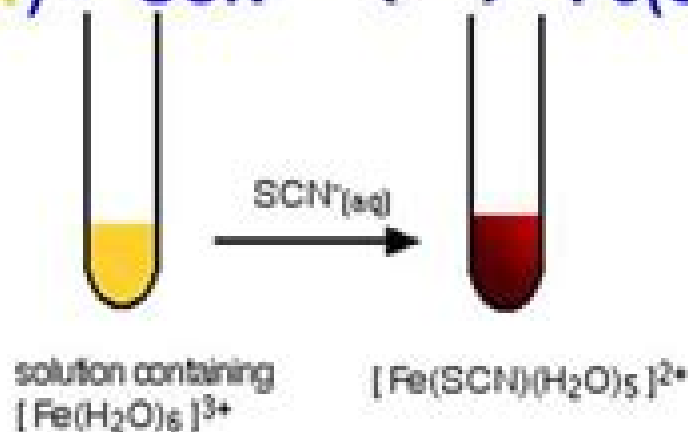


Equilibrium Involving Thiocyanatoiron III Ion



- a) Add KCl (Fe^{3+} forms an insoluble ppt $[\text{FeCl}_3]$ with Cl^{1-})
Result: **Amber Colour**
- b) Add $\text{Fe}(\text{NO}_3)_3$ Possible Stresses: Fe^{+3} and NO_3^-
Result: **Darker Red**
- c) Add KSCN Possible Stresses: K^+ and SCN^-
Result: **Darker Red**
- d) Add NaOH (Fe^{3+} forms ins. ppt $[\text{Fe}(\text{OH})_3]$ with OH^{1-})
Result: **First Orange and then Amber Colour**

Kc Of Iron Thiocyanate

Michael Dub

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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