#### Isolating Copper by Electrolysis

#### I. Abstract

The purpose of this lab is to use electrical energy to isolate copper metal during a chemical reaction. Paper clips hooked up to a six volt battery acted as electrodes to chemically react copper chloride solution into copper and chlorine. This activity is relevant to our class discussions on chemical changes and the law of conservation of mass. Electrolysis is used in many everyday items such as nickel on the five-cent coin, hydrogen from water for fuel in hydrogen powered cars, and the gold plating that covers computer chips in electronic devices.

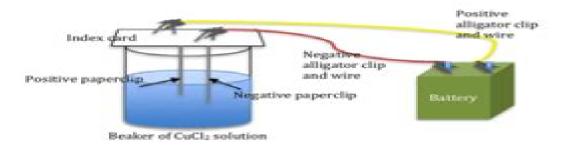
#### II. Hypothesis

I predict that the color of the copper chloride solution will change from blue to copper color.

#### III. Materials

- 250 mL beaker
- 2 paperclips
- six-volt battery.
- index card
- 2 alligator clip wires
- 100 mL of copper chloride solution (CuCl<sub>2</sub>)
- goggles

#### IV. Diagram



#### V. Procedure

- Gather all the materials needed.
- 2. Unbend both paperclips to make a hook shape.
- Push the unbent end of each paperclip through the index card so that they are about three centimeters apart.
- 4. Pour 100 mL of the CuCl2 solution into the beaker.

# **Isolating Copper By Electrolysis Skill Lab Answers**

**K Morrison** 

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Prentice Hall Science Explorer: Teacher's ed ,2005 Industrial Laboratory .1963 **Homogeneous Reactor Project** Oak Ridge National Laboratory, 1956-07 ISEC '88, 1988 The Electrical Journal, 1903 Industrial & Mining The Human Cosmos Jo Marchant, 2021-09-07 Standard, 1911 Scientific American ,1895 The Electrician ,1892 A Best Book of 2020 NPR A Best Book of 2020 The Economist A Top Ten Best Science Book of 2020 Smithsonian A Best Science and Technology Book of 2020 Library Journal A Must Read Book to Escape the Chaos of 2020 Newsweek Starred review Booklist Starred review Publishers Weekly A historically unprecedented disconnect between humanity and the heavens has opened Jo Marchant's book can begin to heal it For at least 20 000 years we have led not just an earthly existence but a cosmic one Celestial cycles drove every aspect of our daily lives Our innate relationship with the stars shaped who we are our art religious beliefs social status scientific advances and even our biology But over the last few centuries we have separated ourselves from the universe that surrounds us It s a disconnect with a dire cost Our relationship to the stars and planets has moved from one of awe wonder and superstition to one where technology is king the cosmos is now explored through data on our screens not by the naked eye observing the natural world Indeed in most countries modern light pollution obscures much of the night sky from view Jo Marchant's spellbinding parade of the ways different cultures celebrated the majesty and mysteries of the night sky is a journey to the most awe inspiring view you can ever see looking up on a clear dark night That experience and the thoughts it has engendered have radically shaped human civilization across millennia The cosmos is the source of our greatest creativity in art in science in life To show us how Jo Marchant takes us to the Hall of the Bulls in the caves at Lascaux in France and to the summer solstice at a 5 000 year old tomb at Newgrange Ireland We discover Chumash cosmology and visit medieval monks grappling with the nature of time and Tahitian sailors navigating by the stars We discover how light reveals the chemical composition of the sun and we are with Einstein as he works out that space and time are one and the same A four billion year old meteor inspires a search for extraterrestrial life The cosmically liberating summary revelation is that star gazing made us human Chemical News and Journal of Physical Science,1912 Engineering ,1909 The Chemical News and Journal of Physical Science, 1912 Amino-acid, Peptide The Chemical & Protein Abstracts ,1974-10 Nuclear Science Abstracts ,1972 Nature Sir Norman Lockyer, 1900 News and Journal of Industrial Science, 1913 Chemical News and Journal of Physical Science William Crookes, James H. Gardiner, Gerald Druce, H. W. Blood-Ryan, 1912 The Chemical News ,1912 The Engineer ,1898 The Electrical **Review** ,1915

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