Balancing Chemical Equations Practice

the elements and atoms on the product and reactant side.

Step 3 – If unbalanced, change coefficients until it's balanced.

Step 1 – Take Inventory of
$$CH_4 + 4CI_2 \rightarrow CCI_4 + 4HCI$$

the elements and atoms
on the product and $C-1 = C-1$
reactant side. $H-4 \neq H-1/4$
Step 2 – Is it balanced? $CI-88 \neq CI-88$

Balanced!

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