

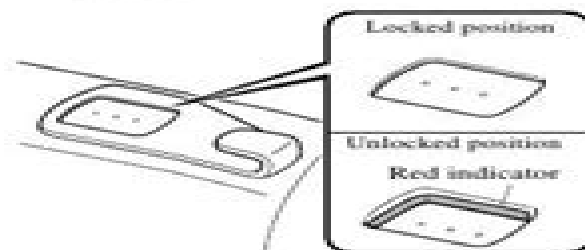
⚠ WARNING

Always make sure the seat belts are fully pulled out from under the seatbacks:

A seat belt caught under a seatback after the seatback is returned to its upright position is dangerous. In a collision or sudden stop, the seat belt cannot provide adequate protection.

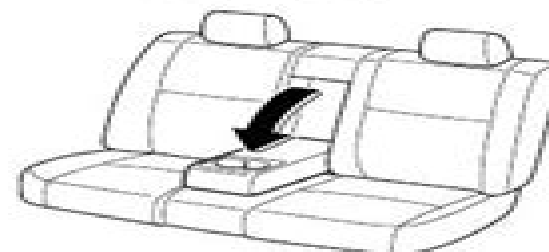
When returning the seatback to the upright position, make sure there is no red indication:

A rear seatback not fully returned and locked in the upright position is dangerous. Sudden stops or maneuvering could cause a seatback to flip forward suddenly resulting in injury. If the red indicator is visible on the back of the rear seatback knob, the seatback is not locked in the upright position.



▼ Armrest *

The rear armrest in the center of the rear seatback can be used (no occupant in the center seat) or placed upright.



* Some models.

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Jiyuan Zhang



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web modeling chemistry 1 u4 ws3 v1 3 compounds of copper and chlorine atoms of cu are heavier than cl atoms by the ratio in compound a a compound a 35 9 g of cl 64 1 g of cu compound b 52 8 g of cl 47 2 g cu a

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web modeling chemistry 1 u3 ws3 v2 0 unit 3 worksheet 3 quantitative energy problems part 1 energy constants $\Delta H_{\text{fus}} 334 \text{ J/g}$
heat of fusion melting or freezing $\Delta H_{\text{vap}} 2260 \text{ J/g}$ heat of vaporization evaporating or condensing $\Delta H_{\text{cond}} 21 \text{ J/g}$ c heat capacity c of
solid water $4.18 \text{ J/g}^\circ\text{C}$ heat capacity c of liquid water

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web chemistry unit 4 worksheet 3 use the following information about the masses of elements in each pair of compounds to help you suggest formulas that account for these ratios 1 compounds of carbon and oxygen compound a 57.1 g C 42.9 g O
compound b 72.7 g C and 27.3 g O determine the value of the ratio mass C : mass O in each