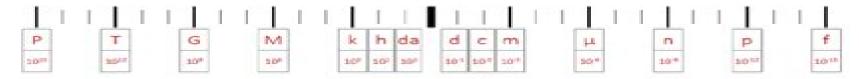
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Science Skills

List the seven fundamental base units and their abbreviations:

	Unit	Abbreviation
Length	Meter	m
Mass	Kilogram	kg
Time	Second	s
Electric Current	Ampere	A
Temperature	Kelvin	K
Amount of Substance	Mole	mol
Luminous Intensity	Candela	cd

Metric Prefixes - List the unit prefixes in their appropriate decimal position



Dimensional Analysis

Convert the following:

20 mi hr¹ → m s¹

$$\frac{20 \ mi}{1 \ hr} \times \frac{1609 \ m}{1 \ mi} \times \frac{1 \ hr}{60 \ min} \times \frac{1 \ min}{60 \ s} = 8.9 \frac{m}{s} = 8.9 \ m \ s^{-1}$$

0.0007 km² → m²

$$0.0007 \, \frac{km^2}{1 \, km} \times \frac{1000 \, m}{1 \, km} \times \frac{1000 \, m}{1 \, km} = 700 \, m^2$$
 or $0.0007 \, \frac{km^2}{1 \, km} \times \left(\frac{1000 \, m}{1 \, km}\right)^2 = 700 \, m^2$

Determine the units for Q:

Q = mc ΔT	m (mass)	kg
$Q = (kg)(J kg^{-1} K^{-1})(K) = \frac{(kg)(J)(K)}{kg K} = J$	c (specific heat)	J kg ⁻¹ K ⁻¹
kg K	ΔT (change in temp)	K

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Michael Bowen-Jones, David Homer

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web in more formal terms chemistry is the study of matter and the changes it can undergo chemists sometimes refer to matter as stuff and indeed so it is matter is anything that has mass and occupies space which is to say anything you can touch or hold common usage might have us believe that chemicals are just those substances in

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