

15. (Pre-lecture) Silver has been used for:
- as a coin metal **(correct)**
  - as a catalyst **(correct)**
16. (Pre-lecture) The atoms of three elements are connected through a covalent bond:
- Carbon, nitrogen, oxygen
  - Hydrogen, carbon, oxygen
  - Carbon, nitrogen, sulfur
  - Hydrogen, carbon, sulfur
17. (Pre-lecture) Which combination of a mixture of  $\text{HNO}_3$  and  $\text{CH}_3\text{COOH}$  is expected to be **soluble** in **water**?
- The mixture of an **acid** and a **base** **(correct)**
  - The mixture of a **strong** and a **weak** acid **(correct)**
  - The mixture of a **strong** and a **weak** base **(correct)**
  - The mixture of a **strong** and a **weak** acid **(correct)**
18. (Pre-lecture) Which of the following is not a **strong** acid?
- $\text{HCl}$
  - $\text{HNO}_3$
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_2\text{O}$

Substance	Is it a strong acid?
$\text{HCl}$	<b>Yes</b>
$\text{HNO}_3$	<b>Yes</b>
$\text{H}_2\text{SO}_4$	<b>Yes</b>
$\text{H}_2\text{O}$	<b>No</b>

19. (Pre-lecture) The following is a list of **strong** acids:  $\text{HCl}$ ,  $\text{HNO}_3$ ,  $\text{H}_2\text{SO}_4$ ,  $\text{H}_2\text{O}$ . Which of the following is **not** a **strong** acid?
- $\text{HCl}$
  - $\text{HNO}_3$
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_2\text{O}$

20. (Pre-lecture) Which of the following is **not** a **strong** acid?
- $\text{HCl}$
  - $\text{HNO}_3$
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_2\text{O}$

$$\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$$

$$\text{HNO}_3 + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{NO}_3^-$$

$$\text{H}_2\text{SO}_4 + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{HSO}_4^-$$

21. (Pre-lecture) Which of the following is **not** a **strong** acid?
- $\text{HCl}$
  - $\text{HNO}_3$
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_2\text{O}$

$$\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{Cl}^-$$

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$$\text{H}_2\text{SO}_4 + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{HSO}_4^-$$

22. (Pre-lecture) Which of the following is **not** a **strong** acid?
- $\text{HCl}$
  - $\text{HNO}_3$
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_2\text{O}$

The following is a list of **strong** acids:  $\text{HCl}$ ,  $\text{HNO}_3$ ,  $\text{H}_2\text{SO}_4$ ,  $\text{H}_2\text{O}$ . Which of the following is **not** a **strong** acid?

- $\text{HCl}$
- $\text{HNO}_3$
- $\text{H}_2\text{SO}_4$
- $\text{H}_2\text{O}$

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**SJ Ball**



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