#### Report FOR EXPERIMENT # 1: DETERMINATION OF MELTING POINTS

Your name TA's name
Your Partner's name Lab Section

#### RESULTS:

Melting point of urea: 134-154 °C

Melting point of trans-cinnamic acid: 133-134 °C

Melting range of a mixture of urea and trans-cinnamic acid: 110-125 °C

Melting point of unknown: 133-133.5 °C

Melting point of unknown mixed with urea: 112-123 °C

Melting point of unknown mixed with Melting point of unknown: 133-133.5 °C

Identity of unknown: Melting point of unknown: 133-133.5 °C

#### CONCLUSION:

In this lab we used melting points to determine the purity of organic compounds. Pure compounds were seen to have a narrow melting range, while impurities were seen to cause the melting range to become lower and broader. From the mixture melting points we were definitively able to identify our unknown as trans-cinnamic acid.

#### QUESTIONS: DETERMINATION OF MELTING POINTS

1. What is a safety hazard associated with this lab session?

One will be using instruments that will be heated to high temperatures. So, one should take precautions when handling these instruments.

# **Lab Report On Melting Points**

**Hygienic Laboratory (U.S.)** 

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Report Kansas State Board of Health,1912 Experimental Organic Chemistry Daniel R. Palleros,2000-02-04 This cutting edge lab manual takes a multiscale approach presenting both micro semi micro and macroscale techniques The manual is easy to navigate with all relevant techniques found as they are needed Cutting edge subjects such as HPLC bioorganic chemistry multistep synthesis and more are presented in a clear and engaging fashion Practical Lab Manual of Pharmaceutical Organic Chemistry - I Dr. Priyanka Gupta Manglik,2024-08-15 Includes a collection of foundational organic chemistry experiments for pharmaceutical studies with step by step procedures observations and safety guidelines

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