

Chapter 1

Answer the following questions:

1. Do you think that Pip's being an orphan affected his life? Why? Give your answer in three sentences at most.
 - Yes, he led a miserable life. He had no one to look after him but his sister who treated him badly.
2. If you were in Pip's place, would you bring the file and food to the stranger? Why?
 - Yes, because I would be afraid of the stranger.
 - No, I would tell my sister and Joe what happened at the marches and ask for their help.
3. Do you think that Pip's older sister led a happy life with her husband? Why?
 - No, because she thought it was bad enough being a blacksmith's wife
4. What do you think would have happened if Pip had told his sister about his behaviour with the convict?
 - She would have punished him for stealing the food and the file.
5. "Suddenly, we heard angry shouts nearby. I was afraid of what was about to happen." What do you think was about to happen?
 - Everyone would know that Pip had helped the convict.
6. Should Pip have told the soldiers about the convict? Why?
 - Yes, because he was a criminal. He should be punished for escaping from prison.
 - No, he shouldn't, because he felt that it was his duty not to tell anyone where he was.

Chapter 2

Answer the following questions:

1. What do you think might have happened if Joe had gone to school?
 - He wouldn't have worked as a blacksmith.
 - He might have become a gentleman.
 - He wouldn't have married Pip's sister.
2. If you were in Pip's place, would you make friends with Joe? Why?
 - Yes, because he treats me well and I trust him.
3. "Then she jumped on me like an eagle on a lamb," said Pip. What did these words reflect about the relationship between Pip and his sister?
 - She doesn't treat her brother well. She had never wished to look after him. Pip probably doesn't like her.
4. The house looked unused and most of the doors were closed. Everything was dark inside. Do you think this atmosphere reflected the circumstances Mrs Havisham had lived in? Why?
 - Yes, as she herself was depressed and never wanted to deal with others after her fiancé had left her on her wedding day.
5. "As I stood before her, I noticed that the clock had stopped at nine." What do you think the stoppage of the clock indicates?
 - It indicates the stoppage of her life itself as she was shocked to discover that her fiancé had left her on her wedding day.
6. Do you think Mrs Havisham was right to let Estella break Pip's heart? Why?

Great Expectations Answers

Emilie Sanchez



Great Expectations Answers:

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web when the hand warmer pack is activated the solid dissolves in water and produces a large temperature change in the form of heat in this laboratory experiment the heat of solution will be calculated from the data for six salts in order to

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web background the hand warmers are filled with sodium acetate click the metal disc and you will notice the liquid changes state and becomes a solid in the process heat is released the sodium acetate is a supersaturated solution

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web oct 9 2021 sodium acetate coin or other small piece of metal plastic bags seal the coin inside a small bag place this bag inside a larger bag containing supersaturated sodium acetate get this by dissolving as much sodium acetate as possible in hot water this chemical hand warmer is reusable

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution ΔH_{soln} or the amount of energy released or absorbed in kJ/mol of the substance being dissolved

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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web feb 2 2017 this is a video outlining a flinn lab on designing a hand warmer it was created by alex brinley charis conwell and siena joy for our ap chemistry class

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web hand warmers provide a unique and fun way to study the chemistry of crystal formation and heat generation by pressing a button in a pouch which contains a supercooled solution you start a rapid exothermic heat producing crystallization in this science fair project you will determine how the starting temperature affects hand warmer chemistry

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web feb 1 2017 hannah boster seth lewis and william makinen s recorded powerpoint presentation summarizing our work on the flinn handwarmer lab created for mr kerns ap

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web lab investigation in today s experiment you will design and execute an experimental procedure to determine which of three ionic compounds is most suitable for use in a hand warmer the three compounds you will use is determined by the group number you are assigned the ideal hand warmer