

15. (Pre-lecture) Study the following reaction:
- a. Write out the **balanced equation**
b. Calculate ΔH_{rxn} **kJ/mol**
16. (Pre-lecture) This reaction is thermodynamically favored at 25°C. Is it also kinetically favored?
a. Explain why/why not.
b. How would you know?
c. How would you know?
d. How would you know?
17. (Pre-lecture) Calculate the standard enthalpy of formation of H_2O and CO_2 from the following data:
- a. H_2O **kJ/mol**
b. CO_2 **kJ/mol**
c. H_2O **kJ/mol**
d. CO_2 **kJ/mol**
18. (Pre-lecture) Calculate the standard enthalpy of formation of H_2O and CO_2 from the following data:
- a. H_2O **kJ/mol**
b. CO_2 **kJ/mol**
c. H_2O **kJ/mol**
d. CO_2 **kJ/mol**

Reaction	Enthalpy of formation (kJ/mol)
H_2O	kJ/mol
CO_2	kJ/mol
H_2O	kJ/mol
CO_2	kJ/mol

19. (Pre-lecture) The following reaction is thermodynamically favored at 25°C. Is it also kinetically favored?
a. Explain why/why not.
b. How would you know?
c. How would you know?
d. How would you know?
20. (Pre-lecture) Calculate the standard enthalpy of formation of H_2O and CO_2 from the following data:
- a. H_2O **kJ/mol**
b. CO_2 **kJ/mol**
c. H_2O **kJ/mol**
d. CO_2 **kJ/mol**

$$\Delta H_{\text{rxn}} = \sum \Delta H_{\text{f}}^{\circ}(\text{products}) - \sum \Delta H_{\text{f}}^{\circ}(\text{reactants})$$

$$\Delta H_{\text{rxn}} = \frac{\text{kJ/mol}}{\text{mol}} = \frac{\text{kJ/mol}}{\text{mol}} = \frac{\text{kJ/mol}}{\text{mol}} = \frac{\text{kJ/mol}}{\text{mol}}$$

21. (Pre-lecture) Calculate the standard enthalpy of formation of H_2O and CO_2 from the following data:
- a. H_2O **kJ/mol**
b. CO_2 **kJ/mol**
c. H_2O **kJ/mol**
d. CO_2 **kJ/mol**

$$\Delta H_{\text{rxn}} = \frac{\text{kJ/mol}}{\text{mol}} = \frac{\text{kJ/mol}}{\text{mol}} = \frac{\text{kJ/mol}}{\text{mol}} = \frac{\text{kJ/mol}}{\text{mol}}$$

22. (Pre-lecture) Calculate the standard enthalpy of formation of H_2O and CO_2 from the following data:
- a. H_2O **kJ/mol**
b. CO_2 **kJ/mol**
c. H_2O **kJ/mol**
d. CO_2 **kJ/mol**

The products of the reaction are H_2O and CO_2 .
Propose a mechanism for the reaction of H_2 and O_2 to form H_2O and CO_2 .
Explain the mechanism of the reaction of H_2 and O_2 to form H_2O and CO_2 .
Explain the mechanism of the reaction of H_2 and O_2 to form H_2O and CO_2 .

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