

Chemistry 20 & Laboratory : Marshall Henderson College Measurement, Measurement Precision

Introduction

This purpose of this lab is to demonstrate how density and percent composition (mass percent) laboratory procedures used for identifying known samples of liquid. In this experiment, you will be using analytical balance as graduated cylinders used to determine mass. Most all procedures are especially good at finding accurate and reproducible measurements, this gives you a good idea of how to use these pieces of equipment and how to use them in your lab.

A measurement is a quantitative observation that compares a mass, volume, or density used in your experiment to a known standard. For example, if you have a sample of a substance and you know its mass and volume, you can find its density. If you have a sample of a substance and you know its mass and volume, you can find its density. If you have a sample of a substance and you know its mass and volume, you can find its density.

The purpose of a measurement is to find out if a sample is a pure substance. This is done by measuring the mass and volume of a sample and comparing it to the known values of a pure substance. If the sample is a pure substance, the mass and volume will be the same as the known values.

Procedure is defined as a series of steps that are used to find out if a sample is a pure substance. The first step is to find out if the sample is a pure substance. This is done by measuring the mass and volume of a sample and comparing it to the known values of a pure substance. If the sample is a pure substance, the mass and volume will be the same as the known values.

In this experiment, you will determine the density of a liquid. In order to do this, you will need to know the mass and volume of the liquid. The mass of the liquid can be found by weighing it on a balance. The volume of the liquid can be found by measuring the volume of the liquid in a graduated cylinder.

It is said that the density of a liquid is a constant.

Since the density of a substance depends on the temperature, it is important to know the temperature of the liquid when you measure its mass and volume.

Materials and Equipment: list of items:

- 100 mL graduated cylinder
- analytical balance
- distilled water (from the lab tap) as a standard
- thermometer

Procedure

The mass of a 100 mL sample of a liquid is 100 g. The mass of a 100 mL sample of a liquid is 100 g. The mass of a 100 mL sample of a liquid is 100 g. The mass of a 100 mL sample of a liquid is 100 g. The mass of a 100 mL sample of a liquid is 100 g.

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**Jerry R Mohrig, Christina Noring
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