

LAB MANUAL WORK INSTRUCTION

Title of Experiment: Reaction of Baking Soda and Vinegar

Objective: To observe the chemical reaction between baking soda (sodium bicarbonate) and vinegar (acetic acid) and to demonstrate the production of carbon dioxide gas.

Pre-Lab Questions

- Write the chemical formula for sodium bicarbonate (baking soda) and acetic acid (vinegar).
- Predict what you think will happen when these two substances are mixed. Why?
- What gas do you expect to be produced from this reaction?

Safety Precautions

- Wear safety goggles and gloves to protect against splashes.
- Perform the experiment in a well-ventilated area to avoid irritation of gases.
- Avoid direct skin contact with vinegar as it is an acid, though mild.

Equipment and Materials

- | | |
|----------------------------------|-------------------|
| 100-ml Beaker | Balloon |
| 10-ml Graduated Cylinder | Funnel |
| Baking Soda (Sodium Bicarbonate) | Safety Goggles |
| Vinegar (Acetic Acid) | Laboratory Gloves |

Procedure

- Place the beaker on a flat surface.
- Using the funnel, fill the balloon with two scoops of baking soda.
- Measure 50 ml of vinegar using the graduated cylinder and pour it into the beaker.
- Carefully fit the neck of the balloon over the mouth of the beaker without letting the baking soda fall into the beaker.
- Once the balloon is securely in place, lift the balloon allowing the baking soda to fall into the vinegar.
- Observe the reaction between the baking soda and vinegar.
- Note the speed and amount of gas produced as indicated by the inflation of the balloon.
- Record any changes in temperature of the beaker.
- Carefully remove the balloon from the beaker.
- Dispose of the vinegar and baking soda solution in the sink and reuse the beaker.

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G. Gregory Haff, Charles Dumke



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Laboratory Manual of Hospital & Clinical Pharmacy Dr. Remeth J. Dias,Dr. Kuldeep U. Bansod,Dr. Vishwajeet S. Ghorpad,Mr. Vijay D. Havaladar,2023-09-29 We are very pleased to put forth the revised edition of Laboratory Manual of Hospital Clinical Pharmacy We have incorporated all the suggestions modified it to make it easier student friendly and relevant in terms of achieving curriculum outcome We are very much thankful to all the learned teachers who have given their feedback whole heartedly We have even incorporated the changes in this manual based on the feedback given by the teachers from all the institutes Now we believe that the manual has been fulfilling the aspirations of Pharmacotherapeutics teachers and students too This manual is prepared as per PCI Education Regulations 2020 for Diploma Course in Pharmacy The methods of all the experiments are reviewed and added from the recent research papers so that the advancement in the

methods or apparatus can be addressed This manual is designed for outcome based education and each experiment is arranged in a uniform way such as practical significance practical outcomes PrOs and its mapping with course outcomes minimum theoretical background resources used procedure precautions observations result conclusion references and related questions Moreover assessment scheme is also given to help the student and teacher to know what to be assessed Every experiment has the component of the activity or role play included so that the students will be able to interact with patients and give them counselling tips on the proper care to be taken in chronic diseases In addition the questions are given at the end of experiments to increase the knowledge of students which would be helpful for them when they will go for higher studies Hope this manual will help the students to learn the concept principles and perform activities and role play counselling the public about diseases and medication We wish you all the best

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