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Label the type of chemical reaction, balance the chemical equation and calculate the Theoretical Yield of Magnesium Oxide based upon your amount of starting material.

Balance the Equation: 2Mg (s) + O2 (s) 2 MgO (s) Reaction Type: Redox

Show all calculations for the theoretical yield of magnesium oxide

molar mass = 80g $2 \times 24 \cdot 48 = <math>\frac{25}{60} = .3125$ 0.0735 = .2352

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