

- 7) If I have 17 liters of gas at a temperature of  $67^{\circ}\text{C}$  and a pressure of 88.89 atm, what will be the pressure of the gas if I raise the temperature to  $94^{\circ}\text{C}$  and decrease the volume to 12 liters?

$$P_1 V_1 T_2 = P_2 V_2 T_1 \quad P_2 = \frac{(88.89 \text{ atm})(17 \text{ L})(367 \text{ K})}{(12 \text{ L})(240 \text{ K})}$$

$$P_2 = \frac{P_1 V_1 T_2}{V_2 T_1}$$

$$P_2 = 192.6 \text{ atm}$$

- 18) I have an unknown volume of gas at a pressure of 0.5 atm and a temperature of 325 K. If I raise the pressure to 1.2 atm, decrease the temperature to 320 K, and measure the final volume to be 48 liters, what was the initial volume of the gas?

$$P_1 V_1 T_2 = P_2 V_2 T_1 \quad V_1 = \frac{(1.2 \text{ atm})(48 \text{ L})(325 \text{ K})}{(0.5 \text{ atm})(320 \text{ K})} = 117 \text{ L}$$

$$V_1 = \frac{P_2 V_2 T_1}{P_1 T_2}$$

- 19) The temperature inside my refrigerator is about  $4^{\circ}\text{C}$ . If I place a balloon in my fridge that initially has a temperature of  $22^{\circ}\text{C}$  and a volume of 0.5 liters, what will be the volume of the balloon when it is fully cooled by my refrigerator?

$$V_2 = ?$$

$$V_1 T_2 = V_2 T_1$$

$$\frac{V_1 T_2}{T_1} = V_2 \quad \frac{.5 \text{ L} \times 277}{295} = .47 \text{ L}$$

- 20) A man heats a balloon in the oven. If the balloon initially has a volume of 0.4 liters and a temperature of  $20^{\circ}\text{C}$ , what will the volume of the balloon be after he heats it to a temperature of  $250^{\circ}\text{C}$ ?

$$V_1 T_2 = V_2 T_1 \quad V_2 = \frac{(0.4 \text{ L})(523 \text{ K})}{(293 \text{ K})}$$

$$V_2 = \frac{V_1 T_2}{T_1}$$

$$V_2 = 0.71 \text{ L}$$

- 21) On hot days, you may have noticed that potato chip bags seem to "inflate", even though they have not been opened. If I have a 250 mL bag at a temperature of  $19^{\circ}\text{C}$ , and I leave it in my car which has a temperature of  $60^{\circ}\text{C}$ , what will the new volume of the bag be?

$$V_1 T_2 = V_2 T_1$$

$$V_2 = \frac{V_1 T_2}{T_1}$$

$$V_2 = \frac{(250 \text{ mL})(333 \text{ K})}{(292 \text{ K})} = 284.1 \text{ L}$$

# Math Skills Specific Heat Answers

**Ralph Thomas Myers**



## **Math Skills Specific Heat Answers:**

Math Skills for the Sciences John G. Pearson, Dennis M. Stone, Richard F. Swindell, 1976      **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction      **Science Spectrum** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03      **Basic Concepts of Chemistry** Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter WileyPLUS sold separately from text      Te HS&T 2007 Shrt Crs M Holt Rinehart & Winston, 2007      **Physics, Volume One: Chapters 1-17** John D. Cutnell, Kenneth W. Johnson, 2014-12-15 Cutnell and Johnson has been the 1 text in the algebra based physics market for almost 20 years The 10th edition brings on new co authors David Young and Shane Stadler both out of LSU The Cutnell offering now includes enhanced features and functionality The authors have been extensively involved in the creation and adaptation of valuable resources for the text This edition includes chapters 1 17      *2024-25 CTET/TET Class VI-VIII Math & Science Solved Papers* YCT Expert Team , 2024 25 CTET TET Class VI VIII Math Science Solved Papers 752 1495 E This book contains the 71 sets of previous year s solved papers with 4262 objective questions      **Waves, Sound, and Light** Glencoe/McGraw-Hill, 2001-06      **Holt Chemistry** Ralph Thomas Myers, 2004      Student Edition McGraw-Hill/Glencoe, 2001-06      **Student Edition Grades 9-12 2018** Cutnell, 2019-03-11      SAT Subject Test Physics Kaplan Test Prep, 2017-01-03 Provides an in depth review of concepts covered on the exam test taking strategies a diagnostic tool and three full length practice tests with detailed answer explanations      **Resources in Education**, 2001-10

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