## **Guided Reading Activity**

netw rks

#### The Ancient Greeks

#### Lesson 3 Classical Greece

#### **Review Questions**

develop their own empire.

- Detail: After an extended war against \_\_\PERSIA\_\_\_\_\_\_, the Greeks, Spartans, and Athenians united to defeat the Persians at \_\_SALAMIS\_\_ and again at \_\_\_PLATAEA\_\_.
- B. Main Idea: The Classical period of Athenian and Greek history lasted from and 461 s.c. to 429 s.c. Historians have called this period the Age of Pericles, in which Athens reached the height of its power and brilliance.
  - Detail: Under Athenian leadership, the \_\_\_DELIAN \_\_ League liberated virtually all
    of the Greek states in the \_\_AEGEAN \_\_\_\_\_ from Persian control.
  - Detail: Pericles created a direct \_\_DEMOCRACY\_\_\_\_\_\_ that included every male citizen and used \_\_OSTRACISM\_\_\_\_\_ to protect against ambitious politicians.
  - Detail: Most people in the ancient world owned \_SLAVES , who worked in the FIELDS , in the home, or on government PUBLIC CONSTRUCTION \_ projects.
  - Detail: Athenian women were \_\_\_\_CITIZENS\_\_\_, but they were excluded from \_\_PUBLIC (political)\_\_ life and expected to work chiefly at \_\_\_home\_\_\_.
- C. Main Idea: After they defeated Persia, the Greek city-states returned to rivalry and war among themselves.
  - Detail: After they defeated the Persians, Athens and the Delian League formed the
     \_\_ATHENIAN\_\_\_\_ Empire, while Sparta and its supporters formed a rival
     PELOPONNESIAN\_\_ League.
  - 2. Detail: The Great \_PELOPONNESIAN WAR\_ between Sparta and \_\_\_\_ATHENS\_\_ weakened the major Greek states and allowed them to fall prey to the growing \_MACEDONIAN\_\_\_\_ empire.

# **History Guided Reading Activity 26 1 Answers**

**TD Snyder** 

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for the nitration  $4\,01\,$  ml of nitric acid  $4\,00\,$  ml of sulfuric acid and  $4\,56\,$  ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only  $4\,56\,$  ml of bromobenzene was used the amount of product predicted to be recovered was  $8\,7533\,$  g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer experiment 11 electrophilic aromatic substitution nitration - Jan 12 2022

on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no2 which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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