

Experiment 1 Lab Report

Determining the Densities of Liquids and Solids

OBJECTIVE:

The objective of Experiment 1 was to find the density of an unknown liquid and an unknown solid.

DISCUSSION:

Density is a standard physical property of a liquid or solid. The density of a subject is expressed as a ratio of the subject's mass to its' volume in the following formula: $\text{Density} = \text{Mass} \div \text{Volume}$. At any specified temperature and pressure the density of a sample is constant. Given the specific density of water at various temperatures and solving the density formula for volume we can accurately determine the volume of our container. The mass of the unknown samples are accurately obtained by weighing and using the measuring by difference method. After obtaining the mass and volume of the unknown sample we can calculate the density of that unknown liquid or solid. By design, Experiment 1 provides the data necessary to calculate the density of the unknown liquid and solid.

EXPERIMENTAL PROCEDURE:

See attached pages.

DATA:

The data collected and calculated that was used to determine the volume of the pycnometer used throughout the experiment is represented in Table 1:

Table 1

	Trial 1	Trial 2
Mass of empty stoppered flask	63.517g	63.515g
Mass of stoppered flask and water	122.194g	122.20g
Mass of water (calculated)	58.677g	58.692g
Temperature of water	20.9°C	21.4°C
Density of water*	0.9980 g/mL	0.9979 g/mL

*Density of water based on temperature, retrieved from – jupiter.plymouth.edu

The data collected and calculated that was used to determine the density of unknown liquid 'B' is represented in Table 2:

Table 2

	Trial 1	Trial 2
Mass of stoppered flask and liquid B	111.590g	111.581g
Mass of liquid (calculated)	48.073g	48.066g
Volume of liquid B (from Table 1 calculation)	58.79mL	58.82mL

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**Cape of Good Hope (South Africa).
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Table of Contents Lab Report Chemistry Matriculation Density Of Water

1. Understanding the eBook Lab Report Chemistry Matriculation Density Of Water
 - The Rise of Digital Reading Lab Report Chemistry Matriculation Density Of Water
 - Advantages of eBooks Over Traditional Books
2. Identifying Lab Report Chemistry Matriculation Density Of Water
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lab Report Chemistry Matriculation Density Of Water
 - User-Friendly Interface
4. Exploring eBook Recommendations from Lab Report Chemistry Matriculation Density Of Water
 - Personalized Recommendations
 - Lab Report Chemistry Matriculation Density Of Water User Reviews and Ratings
 - Lab Report Chemistry Matriculation Density Of Water and Bestseller Lists
5. Accessing Lab Report Chemistry Matriculation Density Of Water Free and Paid eBooks

- Lab Report Chemistry Matriculation Density Of Water Public Domain eBooks
 - Lab Report Chemistry Matriculation Density Of Water eBook Subscription Services
 - Lab Report Chemistry Matriculation Density Of Water Budget-Friendly Options
6. Navigating Lab Report Chemistry Matriculation Density Of Water eBook Formats
 - ePub, PDF, MOBI, and More
 - Lab Report Chemistry Matriculation Density Of Water Compatibility with Devices
 - Lab Report Chemistry Matriculation Density Of Water Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lab Report Chemistry Matriculation Density Of Water
 - Highlighting and Note-Taking Lab Report Chemistry Matriculation Density Of Water
 - Interactive Elements Lab Report Chemistry Matriculation Density Of Water
 8. Staying Engaged with Lab Report Chemistry Matriculation Density Of Water
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lab Report Chemistry Matriculation Density Of Water
 9. Balancing eBooks and Physical Books Lab Report Chemistry Matriculation Density Of Water
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lab Report Chemistry Matriculation Density Of Water
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Lab Report Chemistry Matriculation Density Of Water
 - Setting Reading Goals Lab Report Chemistry Matriculation Density Of Water
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Lab Report Chemistry Matriculation Density Of Water
 - Fact-Checking eBook Content of Lab Report Chemistry Matriculation Density Of Water
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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