

Density Worksheet
Physical Science
 $D = m/V$

Densities of Common Substances @ 20°C				
Substance	Density (g/cm ³)		Substance	Density (g/cm ³)
Oxygen	0.00133		Aluminum	2.70
Hydrogen	0.000084		Iron	7.87
Ethanol	0.785		Copper	8.96
Benzene	0.880		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

1. The ratio of an object's mass to its _____ is called the *density* of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - A. mass = 453 g; volume = 225 cm³
 - B. mass = 5.0 g; volume = 10.0 cm³
 - C. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

Gradpoint Answers Physical Science

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Physical Science Quiz, 1991 *Physical Science Made Easy* Theodore G. Phillips, William Haskel Colburn, 1950

Physical Science Higher Level Thinking Questions Miguel Kagan, Christa Chapman, 1999 Practice Book for Conceptual Physical Science Paul G. Hewitt, John A. Suchocki, Leslie A. Hewitt, 2011-12 This skill building workbook helps students build their confidence and understanding of concepts in the textbook by providing additional questions and activities Answers to all questions are provided at the back of the workbook **Experiences in Physical Science** Laidlaw Brothers, 1983 **Conceptual Physical Science** John Suchocki, Leslie A. Hewitt, 2012 **New Rudman's Questions and Answers on the Graduate Record Advanced Test in Physical Science** Jack Rudman, 1977 **Physical Science Gr. 11** J. M. Lucas, D. Wright, 2007* Physical Science Gr. 10 J. M. Lucas, D. Wright, 2007* **Conceptual Physical Science** Paul G. Hewitt, John A. Suchocki, Leslie A. Hewitt, 2011 This guide provides simple pre class activities and experiments to complement instructors courses Instructions and answers to most of the laboratory questions are provided in the Instructor Manual **Physical Science Concepts in Action** David Frank, Michael Wyssession, Sophia Yancopoulos, 2004-12-01

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