Guided Reading Activity

netw rks

The Ancient Greeks

Lesson 3 Classical Greece

Review Questions

develop their own empire.

- Detail: After an extended war against __\PERSIA______, the Greeks, Spartans, and Athenians united to defeat the Persians at __SALAMIS__ and again at ___PLATAEA__.
- B. Main Idea: The Classical period of Athenian and Greek history lasted from and 461 s.c. to 429 s.c. Historians have called this period the Age of Pericles, in which Athens reached the height of its power and brilliance.
 - Detail: Under Athenian leadership, the ___DELIAN __ League liberated virtually all
 of the Greek states in the __AEGEAN _____ from Persian control.
 - Detail: Pericles created a direct __DEMOCRACY______ that included every male citizen and used __OSTRACISM_____ to protect against ambitious politicians.
 - Detail: Most people in the ancient world owned _SLAVES , who worked in the FIELDS , in the home, or on government PUBLIC CONSTRUCTION _ projects.
 - Detail: Athenian women were ____CITIZENS___, but they were excluded from __PUBLIC (political)__ life and expected to work chiefly at ___home___.
- C. Main Idea: After they defeated Persia, the Greek city-states returned to rivalry and war among themselves.
 - Detail: After they defeated the Persians, Athens and the Delian League formed the
 __ATHENIAN____ Empire, while Sparta and its supporters formed a rival
 PELOPONNESIAN__ League.
 - 2. Detail: The Great _PELOPONNESIAN WAR_ between Sparta and ____ATHENS__ weakened the major Greek states and allowed them to fall prey to the growing _MACEDONIAN____ empire.

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Andrew H. Ryan

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web five of the elements la ce pr nd eu are very reactive and when exposed to air react with oxygen to form an oxide coating that tarnishes the surface for this reason these metals are stored under mineral oil the remainder of the lanthanides are not

as reactive and some gd lu retain their silvery metallic appearance for a long time

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web jun 30 2023 elements lanthanum 57 through lutetium 71 on the periodic table are considered to be lanthanides lanthanides tend to react with oxygen to form oxides the reaction at room temperature can be slow while heat can cause the reaction to

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web to separate the lanthanides from other elements occurring with them they are chemically combined with specific substances to form lanthanide compounds with low solubility oxalates and fluorides for example a process known as ion exchange is then used to separate the lanthanides from each other in this process a solution of the

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used as a general symbol for the lanthanoid elements although lanthanoids scandium sc and yttrium y are sometimes called rare earth elements they are relatively abundant in the earth s crust

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web feb 4 2021 the lanthanides are located on the top row of the two rows of elements found below the main body of the periodic table placing the lanthanides and the actinides the row below the lanthanides below the other elements is just a matter of aesthetics the lanthanides belong to period 6 like cesium barium and hafnium