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web tobacco and other drug use and provided with access to appropriate treatment and support evidence shows that well coordinated and comprehensive support with early access to antenatal care and specialist alcohol or drug treatment can reduce harm and improve outcomes for pregnant women who have problematic alcohol and drug use and their

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web jun 11 2020 introduction antenatal care anc provides monitoring and regular follow up of maternal and fetal health during pregnancy women with appropriate anc tend to have better delivery and birth outcomes methods baseline cross sectional data from pregnant women who participated in the mutaba ah mother and child health study between may

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web jan 2 2018 methadone maintenance enables stabilisation of the mother s drug use and lifestyle and can also facilitate access to comprehensive antenatal and postnatal care furthermore by reducing or eliminating illicit drug use it can help to stabilise the in utero environment while not increasing the risk of congenital abnormalities in the foetus

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web oct 1 2014 type guideline date of publication 01 october 2014 author centre for alcohol and other drugs these clinical guidelines are intended to support a range of health care workers who care for pregnant and breastfeeding women with substance use issues and their infants and families

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- \text{aq} + 2\text{Ag}^+ \text{aq}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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1 $\text{K}_3\text{PO}_4 \text{aq}$ + 1 $\text{Al}^{3+} \text{aq}$ + 3 $\text{OH}^- \text{aq}$ + 3 $\text{H}_2\text{O} \text{l}$ ionic $3\text{Cl}^- \text{aq}$ + 1 $\text{Ag}^+ \text{aq}$ + 3 $\text{NH}_4^+ \text{aq}$ + 3 $\text{K}^+ \text{aq}$ net ionic $\text{AgCl} \text{s}$ + 2 $\text{H}_2\text{O} \text{l}$ + 3 $\text{mg} \text{s}$

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oct 2 2019 the net ionic equation is a chemical equation for a reaction that lists only those species participating in the reaction the net ionic equation is commonly used in acid base

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq 2f aq because two nh₄ aq

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